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## Synthetic Study of Marine Macrolide Swinholide A. Stereocontrolled Synthesis of the C11 - C23 Segment

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Abstract: The C11-C23 segment 4 of swinholide A (1) was synthesized stereoselectively starting from (S)-malic acid via double nitroaldol reaction under high pressure and differentiation of the left and right parts of the symmetric compound 10 as the key steps.

Swinholide A (1),<sup>1</sup> isolated from the marine sponge *Theonella swinhoei*, exhibits potent cytotoxic activity and the unique structure was finally established by Kitagawa *et al.*<sup>2</sup> as a 44-membered dimeric macrolide. Recently, pre-swinholide A (2), a monomeric seco acid of 1, was also isolated from *Theonella swinhoei*<sup>3</sup> and the first total synthesis of 2 has been reported by Paterson *et al.*<sup>4</sup> We have also engaged in the synthesis of these compounds<sup>5</sup> and now report the stereocontrolled synthesis of the C11-C23 segment 4.

Retrosynthetic analysis of 1 (and 2) reveals that the polyol part 3 corresponding to the C11-C32 segment could be synthesized stereoselectively via aldol reaction of segments 4 and 5.6 Segment 4 would be synthesized from the C<sub>2</sub> symmetric ketone 6 prepared based on our method for the convergent synthesis of 1,3-polyol.<sup>7</sup>



The optically active aldehyde  $7^7$  prepared from (S)-malic acid was used as a starting material for the synthesis of 4. Addition of nitromethane to 2 equiv of aldehyde 7 in Et3N under high pressure (5.5 kbar, 5

h) caused double nitroaldol reaction effectively producing a mixture of diols 8 in 90% yield.<sup>8</sup> Deprotection of the THP group in 8, and protection of the resulting tetrol as isopropylidene acetal followed by KMnO4 oxidation of the nitro group gave a mixture of three ketones 9 due to the isomers at the C15 and C17 positions. On treatment of the mixture 9 with K<sub>2</sub>CO<sub>3</sub> in MeOH, epimerization took place producing the  $C_2$  symmetric ketone 6 (<sup>1</sup>H NMR:  $\delta$  4.69 (dd, J=3.0, 12.0 Hz; C15-and C17-H)), exclusively.<sup>7</sup>



(a) MeNO<sub>2</sub> (1 equiv), aldehyde 7 (2 equiv), Et<sub>3</sub>N, at 5.5 kbar, rt (90%); (b) aq AcOH, rt; (c) Me<sub>2</sub>C(OMe)<sub>2</sub>, CSA, acetone, rt; (d) ⊁BuONa, KMnO<sub>4</sub>, MgSO<sub>4</sub>, H<sub>2</sub>O, PhH, rt; (e) K<sub>2</sub>CO<sub>3</sub>, MeOH, rt (54% from 8); (f) Ph<sub>3</sub>P\*Mel<sup>-</sup>, *n*-BuLi, THF, rt (98%); (g) BH<sub>3</sub>-Me<sub>2</sub>S, THF, rt; 30%H<sub>2</sub>O<sub>2</sub>, 15% NaOH, rt.

Next, we examined the introduction of methyl group to the C16 position. After introduction of exomethylene group by Wittig reaction, the resulting 10 was subjected to hydroboration with BH3·Me2S (1 equiv) giving the alcohol 12 (42%; 84% based on the consumed 10) and the partially reduced product 13 (8%). On treatment of 10 with an excess of BH3·Me2S (5 equiv), this single isopropyl ether 13 was obtained as the major product in 48 % yield along with 12 (24%).<sup>9</sup> Formation of 13 shows that hydroboration of the exomethylene group and differential reductive cleavage of the isopropylidene acetal took place simultaneously in the present reaction, which was expected to enable the complete differentiation of the right and left parts of the C<sub>2</sub> symmetric 10. Thus, after replacement of 14 with an excess of BH3·Me2S gave, as expected, the fully differentiated diol 15 as a single diastereomer in 88% yield. The stereochemistry of 15 was confirmed based on the NOE measurement of oxetane 16 prepared from 15; NOE was observed between C15-H and C17-H.

Conformational analyses of the probable intermediate 11 in the present reaction were undertaken. MM2 calculations<sup>10</sup> of 11 (CH<sub>2</sub>BR<sub>2</sub> and CH<sub>2</sub>CH<sub>2</sub>OBn were replaced by CH<sub>2</sub>OH and Me, respectively) revealed that the most stable conformation was i; the energy difference between i and an alternative conformation ii leading to the other diastereomer was ca. 1.5 kcal/mol. Therefore, the present reaction would proceed via the transition state shown in Fig 1, in which boron atom can coordinate with only one oxygen atom (C15-O) of two acetals producing exclusively the MPM ether 15 by the successive reductive cleavage of the acetal ring.



Diol 15 was then converted into 17 in three steps; 1) tosylation of the primary alcohol, 2) LiAlH4 reduction, 3) O-methylation with MeI-KH. Deprotection of the benzyl groups in 17 with Raney Ni, acid hydrolysis of the benzylidene group, and treatment with DDQ produced triol 18. After oxidation of 18 with Ag<sub>2</sub>CO<sub>3</sub>-Celite, the resulting  $\beta$ -hydroxy- $\delta$ -lactone 19 was methylated with LDA and MeI giving 20



(a) p-TsCl, py, rt; (b) LiAlH<sub>4</sub>, Et<sub>2</sub>O, 0°C; (c) KH, MeI, THF, 0°C (76% from 15); (d) H<sub>2</sub>, Raney Ni, EtOH, rt; (e) aq AcOH, rt; (f) DDQ, CH<sub>2</sub>Cl<sub>2</sub>, rt (61% from 17); (g) Ag<sub>2</sub>CO<sub>3</sub>-Celite, PhH, reflux (51%); (h) LDA, MeI, HMPA, THF, -78 ~ -40°C (71%; 83% based on the consumed 19); (l) LIBH<sub>4</sub>, THF, rt; (j) *t*-BuPh<sub>2</sub>SiCl, imidazole, DMF, rt; (k) Me<sub>2</sub>C(OMe)<sub>2</sub>, PPTS, CH<sub>2</sub>Cl<sub>2</sub>, rt; (l) *n*-Bu<sub>4</sub>NF, THF, rt; (m) (COCl)<sub>2</sub>, DMSO, CH<sub>2</sub>Cl<sub>2</sub>, -78°C; Et<sub>3</sub>N, -78 ~ 0°C (61% from 20); (n) imide 22, *n*-Bu<sub>2</sub>BOTf, *i* Pr<sub>2</sub>NEt, CH<sub>2</sub>Cl<sub>2</sub>, -78°C ~ rt (88%; 95% based on the consumed 21); (o) 30% H<sub>2</sub>O<sub>2</sub>, LiOH, aq THF, 0°C; (p) CH<sub>2</sub>N<sub>2</sub>, Et<sub>2</sub>O, rt; (q) Et<sub>3</sub>SiOTf, 2,6-lutidine, CH<sub>2</sub>Cl<sub>2</sub>, 0°C; (r) DIBAH, CH<sub>2</sub>Cl<sub>2</sub>, -78°C; (s) PDC, MS3A, CH<sub>2</sub>Cl<sub>2</sub>, rt (56% from 23).

stereoselectively.<sup>11</sup> δ-Lactone 20 was converted into aldehyde 21 in five steps; 1) LiBH4 reduction, 2) silulation of the primary alcohol, 3) acetonization of diol, 4) deprotection of silul ether, 5) Swern oxidation. The aldehyde 21 was subjected to aldol reaction with optically active imide 22 producing 21,22-syn-aldol adduct 23 stereoselectively.<sup>12,13</sup> Finally, the aldol 23 was successfully converted into the desired aldehyde 4,14 corresponding to the C11-C23 segment, in five steps; 1) hydrolysis of imide with LiOOH, 2) esterification, 3) silulation of C21-hydroxyl group, 4) DIBAH reduction, 5) PDC oxidation.

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- This reaction at atmospheric pressure (rt, 18 hr) gave 8 in only 15 % yield. High pressure reaction was carried out by using the apparatus made by Instrumentation and Characterization Center of this institute (RIKEN). The RIKEN apparatus is capable of holding a 150 ml volume at 15 kbar or 50 ml 8 volume at 20 kbar.
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- 10. Molecular modeling studies were carried out with the MacroModel program using MM2\* force field (MacroModel V4.0). In the case of 11 (CH2BR2 and CH2CH2OBn were replaced by Et and Me, respectively), the most stable conformation was also the similar one as i.
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- 12.
- The stereochemistry of aldol adduct 23 was unequivocally confirmed based on the <sup>1</sup>H NMR analysis of 24 prepared from 4 in two steps as shown below. <sup>1</sup>H NMR (500 MHZ; CDCl<sub>3</sub>):  $\delta$  4.71 (dd, J<sub>20,21</sub>= 13 4.9 Hz, J21,22=11.3 Hz, C21-H), 5.30 (d, J22,23=9.2 Hz, C23-H).



14. Data for 4: IR (neat): 1730, 1615, 1520 cm<sup>-1</sup>,  $[\alpha]_D$  -0.8° (c 1.75, CHCl<sub>3</sub>), <sup>1</sup>H NMR (500 MHz; CDCl<sub>3</sub>):  $\delta$  0.83, 0.90, 1.11 (each d, J=7.1 Hz; Me x 3), 3.35 (s; OMe), 3.80 (s; Ar-OMe), 4.27 (dd, J=4.1, 11.3 Hz; O-CH), 4.32 (dd, J=2.8, 7.1 Hz; O-CH), 5.48 (s; O-CH-O), 9.72 (s; CHO).

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